**Total and Net Ionic Equations Practice Problems**

Write the balanced formula equation, total ionic equation and net ionic equation for each of the following:

1. potassium oxide (aq) + strontium chloride(aq) 🡪

2. sodium phosphate (aq) + cobalt(II) chloride (aq) 🡪

3. chloric acid (aq) + magnesium hydroxide (aq) 🡪

4. potassium iodide (aq) + nickel (II) chloride (aq) 🡪

5. silver nitrate (aq) + sodium oxalate (aq) 🡪

6. sodium sulfide(aq) + nitric acid(aq) 🡪